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1. PRODUCT IDENTIFICATION

Product Name LEAD OXIDE

Other Names PbO, lead oxide (PbO), lead containing, lead monoxide, Litharge

Use Lead oxide – a raw material for the production of "lead paste" for battery manufacture

Supplier Name and Address Century Yuasa Batteries

37-65 Cobalt St Carole Park QLD 4300

Telephone (07) 3361 6161 **Emergency (24 Hours)** (07) 3361 6707

Relevant identified uses Raw material for inorganic and organic lead salt manufacture, e.g. battery oxides,

2. HAZARD(S) IDENTIFICATION

HAZARDOUS CHEMICAL DANGEROUS GOODS. According to the Model WHS Regulations and the ADG Code.

Poisons Schedule S6 Classified as S6:- Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP)

Signal Word DANGER

GHS Classification Acute Toxicity (Oral) Category 4, Acute Toxicity (Inhalation) Category 4, Reproductive Toxicity Category 1B, STOT - RE

Category 2, Acute Aquatic Hazard Category 1, Chronic Aquatic Hazard Category 1

GHS Label Elements



protection





	Harmful	Health Hazard	Enviror	nment	
Hazard Statements	H302	Harmful if swallowed		H373	May cause damage to organs through prolonged or repeated exposure
	H332	Harmful if inhaled		H400	Very toxic to aquatic life
	H360	May damage fertility or the unbo child	orn	H410	Very toxic to aquatic life with long lasting effects
Precautionary Statements	<u>Prevention</u>			Response	
	P101	If medical advice is needed, have product container or label at han		P308+P313	IF EXPOSED: or concerned: Get medical advice / attention.
	P102	Keep out of reach of children		P314	Get medical advice / attention if you feel unwell.
	P103	Read label before use.		P301+P312	IF SWALLOWED: Call a POISON CENTER / doctor / physician / first aider / if you feel unwell.
	P201	Obtain special instructions before use.	re	P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.
	P260	Do not breathe dust / fume / gas mist / vapours / spray.	s /	P391	Collect spillage.
	P270	Do not eat, drink or smoke when using this product.	1	<u>Storage</u>	
	P271	Use only outdoors or in a well-ventilated area.		P405	Store locked up.
	P273	Avoid release to the environmen	nt	<u>Disposal</u>	
	P280	Wear protective gloves / protectic clothing / eye protection / face	ive	P501	Dispose of contents, container to authorised chemical landfill or if organic, to

high temperature incineration



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3. COMPOSITION, INFORMATION ON INGREDIENTS

Ingredient	Identification	Content % weight
Lead Monoxide (PbO)	CAS 1317-36-8	85%
Lead Oxide (PbO ₂)	CAS 1309-60-0	15%

4. FIRST AID MEASURES

DESCRIPTION OF FIRST AID MEASURES

Eye contact

If this product comes in contact with the eyes:

- Wash out immediately with fresh running water.
- Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by
 occasionally lifting the upper and lower lids.
- Seek medical attention without delay; if pain persists or recurs seek medical attention.
- Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

Skin contact

If skin contact occurs:

- Immediately remove all contaminated clothing, including footwear.
- Flush skin and hair with running water (and soap if available).
- Seek medical attention in event of irritation.

Inhalation

If fumes or combustion products are inhaled:

- Remove from contaminated area.
- Lay patient down. Keep warm and rested.
- Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.
- Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.
- Transport to hospital, or doctor.

Ingestion

- IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.
- For advice, contact a Poisons Information Centre or a doctor.
- Urgent hospital treatment is likely to be needed.
- In the meantime, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient's condition.
- If the services of a medical officer or medical doctor are readily available, the patient should be placed in his / her care and a copy of the SDS should be provided. Further action will be the responsibility of the medical specialist.
- If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the SDS.
- Where medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructed otherwise:
- INDUCE vomiting with fingers down the back of the throat, ONLY IF CONSCIOUS. Lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.
- NOTE: Wear a protective glove when inducing vomiting by mechanical means.

MEDICAL ATTENTION AND SPECIAL TREATMENT. Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

- Gastric acids solubilise lead and its salts and lead absorption occurs in the small bowel.
- Particles of less than 1 um diameter are substantially absorbed by the alveoli following inhalation.
- Lead is distributed to the red blood cells and has a half-life of 35 days. It is subsequently redistributed to soft tissue & bone-stores or eliminated. The kidney accounts for 75% of daily lead loss; integumentary and alimentary losses account for the remainder.
- Neurasthenic symptoms are the most common symptoms of intoxication. Lead toxicity produces a classic motor neuropathy. Acute encephalopathy appears infrequently in adults. Diazepam is the best drug for seizures.
- Whole-blood lead is the best measure of recent exposure; free erythrocyte protoporphyrin (FEP) provides the
 best screening for chronic exposure. Obvious clinical symptoms occur in adults when whole-blood lead exceeds
 80 ug / dL.
- British anti-lewisite is an effective antidote and enhances faecal and urinary excretion of lead. The onset of action of BAL is about 30 minutes and most of the chelated metal complex is excreted in 4-6 hours, primarily in the bile. Adverse reaction appears in up to 50% of patients given BAL in doses exceeding 5 mg / kg. CaNa2EDTA has also been used alone or in concert with BAL as an antidote. D-penicillamine is the usual oral agent for mobilisation of bone lead; its use in the treatment of lead poisoning remains investigational. 2,3-dimercapto-1-propanesulphonic acid (DMPS) and dimercaptosuccinic acid (DMSA) are water soluble analogues of BAL and their effectiveness is undergoing review. As a rule, stop BAL if lead decreases below 50 ug / dL; stop; CaNa2EDTA if blood lead decreases below 40 ug / dL or urinary lead drops below 2 mg / 24hrs.



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5. FIRE FIGHTING MEASURES

Recommended Extinguishing Media











Water spray or fog.

Foam

Dry chemical powder.

Carbon dioxide.

x

BCF\ Vaporising Liquid (Where regulations permit).

Extinguishing Media Incompatibilities

- There is no restriction on the type of extinguisher which may be used.
- · Use extinguishing media suitable for surrounding area.

Specific Hazards Hazardous Decomposition

- Non-combustible.
- Not considered a significant fire risk, however containers may burn. Decomposition may produce toxic fumes of metal oxides, which may emit poisonous fumes

Fire Incompatibility

None known.

Fire Fighting, Special Protective Equipment & Precautions

- Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves.
- Prevent, by any means available, spillage from entering drains or water courses.
- Use fire fighting procedures suitable for surrounding area.
- DO NOT approach containers suspected to be hot.
- Cool fire exposed containers with water spray from a protected location.
- If safe to do so, remove containers from path of fire.
- Equipment should be thoroughly decontaminated after use.

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions

Environmental

Precautions

- Avoid contact with skin and eyes.
- Prevent, by any means available, spillage from entering drains or water course.

Methods and materials of containment and cleaning up

- With a clean shovel, transfer spilled material into clean-labelled containers for disposal.
- Prevent from entering drains, sewers, streams or other bodies of water. If contamination of sewers or waterways has occurred, advise the local emergency services

Protective Equipment

Personal Protective Equipment advice is contained in Section 8 of the SDS.

Emergency Procedures Minor Spills

- Oleana all and
- Clean up all spills immediately.
- Avoid contact with skin and eyes.
- Control personal contact with the substance, by using protective equipment.
- Use dry clean up procedures and avoid generating dust.
- Place in a suitable, labelled container for waste disposal.

Major Spills

- Remove all ignition sources.
- · Clean up all spills immediately.
- · Avoid contact with skin and eyes.
- Control personal contact with the substance, by using protective equipment.
- Clear area of personnel and move upwind.
- Alert Fire Brigade and tell them location and nature of hazard.
- · Wear breathing apparatus plus protective gloves.
- · Prevent, by any means available, spillage from entering drains or water courses.
- Stop leak if safe to do so.
- · Contain or absorb spill with sand, earth or vermiculite.
- Collect recoverable product into labelled containers for recycling.
- Collect solid residues and seal in labelled drums for disposal.
- Wash area and prevent runoff into drains.
- After clean-up operations, decontaminate and launder all protective clothing and equipment before storing and reusing.
- If contamination of drains or waterways occurs, advise emergency services, other related regulatory authorities such
 as environmental protection and local council.

7. HANDLING AND STORAGE

Safe Handling

- · Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- DO NOT allow material to contact humans, exposed food or food utensils.
- When handling, DO NOT eats, drink or smoke.
- Keep containers securely sealed when not in use.
- Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately. Launder contaminated clothing before re-use.
- Use good occupational work practice.



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Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions

Storage

- Store in original containers.
- Keep containers securely sealed.
- Store in a cool, dry, well-ventilated area.
- Store away from incompatible materials and foodstuff containers.
- Protect containers against physical damage and check regularly for leaks.

Suitable container

- Packing of Lead oxide product in light weight metal or plastic packages may result in container collapse with product
- Lined metal can, lined metal pail / can.
- Plastic pail.
- Poly-lined drum.
- Check all containers are clearly labelled and free from leaks.
- For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):
- Removable head packaging;

Storage incompatibility

Is incompatible with aluminium carbide, barium sulphide, silicon, sulphuryl chloride, hydrogen peroxide, chemical active metals, aluminium, combustible materials, lithium carbide, chlorinated rubber, chlorine, boron, hydrides, ethylene, fluorine, sulphides, acetylides and strong reducing agents.





















FLAMMABLES

EXPLOSIVES

ACUTE TOXIC

OXIDISERS

HARMFUL

IRRITANT

CORROSIVE

EXPOSURE CONTROLS, PERSONAL PROTECTION

AUSTRALIAN EXPOSURE STANDARDS (Occupational Exposure Limits)

Ingredient	Material name	TWA	STEL
Lead Monoxide (PbO)	Lead, inorganic dusts & fumes (as Pb)	0.15 mg / m3	Not Available
Lead oxide (PbO ₂)	Lead, inorganic dusts & fumes (as Pb)	0.15 mg / m3	Not Available

APPROPRIATE ENGINEERING CONTROLS

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

- Process controls which involve changing the way a job activity or process is done to reduce the risk.
- Enclosure and / or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

PERSONAL PROTECTION



Respirator Type

Where the concentration of gas / particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Type E-P Filter of sufficient capacity.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	P1 Air-line*	-	PAPR-P1 -
up to 50 x ES	Air-line**	P2	PAPR-P2
up to 100 x ES	-	P3	•
		Air-line*	-
100+ x ES	-	Air-line**	PAPR-P3

Negative pressure demand

Other Protection

- Eyewash unit.
- Barrier cream.
- Skin cleansing cream.



Eye Protection

- Safety glasses with side shields Chemical goggles.
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants.



Glove Type

Wear chemical protective gloves, e.g. PVC



Clothing

Overalls.



Foot wear

Wear safety footwear or safety gumboots e.g. Rubber

PHYSICAL AND CHEMICAL PROPERTIES

Pale yellow / bright yellow orange coloured, very dense, odourless powder. Does not mix with water but Appearance classed as "soluble" under the provisions of SP 199 of the UN Transport Code.

Odour Not Available Vapour pressure (kPa) 1 mm Hg @ 973 °C Odour threshold Not Available Vapour density (Air = 1) Not Applicable

^{**} Continuous flow



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Immiscible

Not Available

рΗ Relative density (Water = 1) Not Applicable 888 °C Melting point / freezing point (°C) Solubility in water (g,L) Initial boiling point and boiling range 1472 °C Partition coefficient: n-octanol / water

Not Applicable

Flash point Not Applicable Molecular weight (g / mol) 223.2 g / mol

Evaporation rate Decomposition temperature (°C) >500-700 °C lead fumes given off Not Available

Flammability Not Applicable Viscosity Not Available

Upper, lower flammability or

explosive limits

10. STABILITY AND REACTIVITY

Reactivity

See section 7 and this section under Chemical stability

- Lead oxide:- is a strong oxidiser
- Attacks some plastics, rubber and coatings

Possibility of hazardous reactions

See section 5 & 7

- Reacts explosively with 90% performic acid, rubidium acetylide
- Reacts violently with strong oxidisers,
- Reacts violently with aluminium, sodium, zirconium, titanium, boron or silicon, when heated forms impact sensitive explosive mixtures with dichloromethylsilane

Incompatible materials

See section 7

Is incompatible with aluminium carbide, barium sulphide, silicon, sulphuryl chloride, hydrogen peroxide, chemical active metals, aluminium, combustible materials, lithium carbide, chlorinated rubber, chlorine, boron, hydrides, ethylene, fluorine, sulphides, acetylides and strong reducing agents.

Chemical stability

- Product is considered stable
- Hazardous polymerisation will not occur.
- Unstable in the presence of incompatible materials

Hazardous decomposition products See section 5

Thermal decomposition may produce oxides of lead.

11. TOXICOLOGICAL INFORMATION ACUTE EFFECTS

No adverse health effects expected if the product is handled in accordance with this safety Data sheet and the product Label. Symptoms or effects that may arise if the product is mishandled and overexposure occurs are:-

Inhaled

The material is not thought to produce either adverse health effects or irritation of the respiratory tract following inhalation (as classified by EC Directives using animal models). Nevertheless inhalation of dusts, or fumes, especially for prolonged periods, may produce respiratory discomfort and occasionally, distress. Adverse systemic effects have been produced following exposure of animals by at least one other route and good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting. Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.

Inhalation of dusts, generated by the material, during the course of normal handling, may be harmful.

If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further risk if handling and use of the material result in excessive exposures.

Ingestion

Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.

Skin contact

The material is not thought to be a skin irritant (as classified by EC Directives using animal models). Abrasive damage however, may result from prolonged exposures.

Skin contact with the material may damage the health of the individual; systemic effects may result following

Open cuts, abraded or irritated skin should not be exposed to this material

Entry into the blood-stream, through for example, cuts, abrasions or lesions, may produce systemic injury with harmful

Eye

Although the material is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may cause transient discomfort characterised by tearing or conjuctival redness (as with windburn). Slight abrasive damage may also result.

Chronic effects

An inorganic compound such as Lead is a cumulative harmful poison when exposed in small amounts can raise the body's content to toxic levels. Prolonged or repeated exposure to lead toxicity effects the nervous system (memory loss, tiredness, headaches, fatigue, irritability, decreased libido, dizziness, depression, encephalopathy (brain damage caused by altered brain function and structure), behavioural effects, altered mood states, disturbances in hand-eye coordination, reaction times, visual motor performance, and mental performance, disturbances to vision, changes in hearing, muscle and joint weakness of the arms and legs, (foot-drop and wrist-drop), heart / blood vessels (reduced haemoglobin synthesis and production, reduced life span and function of red blood cells, anaemia, increased blood pressure), digestive system (loss of appetite, anorexia, with severe abdominal pain, diarrhoea, inflammation of the stomach walls (gastritis) and colic, cramps, nausea, vomiting, constipation, weight loss and decreased urination, deposition of blue lead-line on the gums), kidneys / urinary system (reversible / irreversible kidney damage) and



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endocrine system. Increased levels of lead result in increased brain damage, coma and death in extreme cases.

- Long term exposure to high dust concentrations may cause changes in lung function i.e. pneumoconiosis; caused by particles less than 0.5 micron penetrating and remaining in the lung.
- Ample evidence from experiments exists that there is a suspicion this material directly reduces fertility.
- Lead can cross the placenta, and cause miscarriage, stillbirths and birth defects. Exposure before birth can cause mental retardation, behavioural disorders and infant death.
- Exposure to the material for prolonged periods may cause physical defects in the developing embryo (teratogenesis).
- Ample evidence exists that developmental disorders are directly caused by human exposure to the material.
- Lead can accumulate in the skeleton for a very long time.

Acute Toxicity	Skin Irritation / Corrosion	Serious Eye Damage / Irritation	Respiratory Or Skin Sensitisation	Mutagenicity	Carcinogenicity	Reproductivity	Stot - Single Exposure	Stot - Repeated Exposure	Aspiration Hazard
✓	•	•	•	1	1	✓	1	✓	(

✓ = Data required to make classification available ×= Data available but does not fill the criteria for classification

①= Data Not Available to make classification

12. ECOLOGICAL INFORMATION

Toxicity

- DO NOT discharge into sewer or waterways.
- Very toxic to aquatic organisms. May cause long-term adverse effects in the aquatic environment.
- Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.
- Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

- Environmental Fate: Lead is assessed as low hazard if it remains in its solid, massive, metallic form. Lead, in the form of alkyls, has been introduced to the environment primarily from leaded gasoline / petrol. These are converted to water-soluble lead compounds of high toxicity and availability to plants.
- Atmospheric Fate: Lead is primarily an atmospheric pollutant that enters soil and water as fallout, a process determined by the physical form involved and particle size. Lead, in the form of alkyls, has been introduced to the environment primarily from leaded gasoline / petrol. Lead is absorbed by mammals / humans via vapors, contaminated dust, and fumes.
- Terrestrial Fate: Soil Lead alkyls easily leach from soil to contaminate water sources close to highways. Plants -Lead alkyls that have been converted to water soluble lead compounds have high toxicity / availability to plants.
- Aquatic Fate: Lead that has entered the aquatic system is expected to be found in sediments.
- Ecotoxicity: Soluble or insoluble lead may enter the environment and accumulate. Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment

Fish

The following applies to lead compounds in **Daphnia** general: fish: lethal from 1.4 mg / I up S. gairdnerii: LC50: 0.14 mg / I / 96h L. idus LC50: 546 mg / I fish test LC50: 236 mg / I (calc. as free lead).

Algae

The following applies to lead compounds in Bacteria general: algae: Sc. quadricauda toxic from 3.7 mg/l up M. aeruginosa 0.45 mg/l

(calc. as free lead).

Other Organisms

The following applies to lead compounds in general: protozoa: E. sulcatum toxic from 0.02 mg / I up U. parduczi toxic from 0.07 mg / I up (calc. as free lead).

Degradability

No Data available for all ingredients

Bio-accumulative Potential

Lead Monoxide LOW (BCF = 43)

Mobility in Soil No Data available for all ingredients

Other Adverse Effects

No Data available for all ingredients

13. DISPOSAL CONSIDERATIONS

Safe Handling & Disposal

- Dispose in accordance with federal, state or local regulations.
- Disposal of
- Contaminated Packaging .
- Containers may still present a chemical hazard / danger when empty.
 - Return to supplier for reuse / recycling if possible.
 - Otherwise:
 - If container cannot be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
 - Where possible retain label warnings and SDS and observe all notices pertaining to the product.
 - Legislation addressing waste disposal requirements may differ by country, state and / or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.
 - This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. In most instances the

The following applies to lead compounds in general: fish: lethal from 1.4 mg / l up S. gairdnerii: LC50: 0.14 mg / I / 96h L. idus LC50: 546 mg / I fish test LC50: 236 mg / I (calc. as free lead).

The following applies to lead compounds in general: algae: Sc. quadricauda toxic from 3.7 mg / I up M. aeruginosa 0.45 mg / I (calc. as free lead).



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supplier of the material should be consulted.

- DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
- Recycle wherever possible or consult manufacturer for recycling options.
- Consult State Land Waste Management Authority for disposal.
- Bury residue in an authorised landfill.
- Recycle containers if possible, or dispose of in an authorised landfill.
- Observe all label safeguards until containers are cleaned and destroyed.

Environmental Regulations

Dispose in accordance with federal, state or local regulations.

Regulations • Refer to section 15

14. TRANSPORT INFORMATION

REGULATED FOR TRANSPORT OF DANGEROUS GOODS ADG

UN Number 2291

Proper Shipping Name LEAD COMPOUND, SOLUBLE, N.O.S.

Transport Hazard Class Class: 6.1 Sub risk: Not Applicable

Packing group III

Environmental Hazards No relevant data

Special Precautions Special provisions 199, 274

Limited quantity 5 kg

Additional Information Marine Pollutant: Yes

Hazchem Code 2Z





15. REGULATORY INFORMATION

SAFETY, HEALTH AND ENVIRONMENTAL REGULATIONS, LEGISLATION

Lead Monoxide (PbO) Lead Oxide (PbO2) are found on the following

regulatory lists

Australia Exposure Standards", "Australia Inventory of Chemical Substances (AICS)", "International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs", "International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft",

"Australia Hazardous Substances Information System - Consolidated Lists"

Poisons Schedule (Australia) 6 AICS Status All the constituents of this product are listed

APVMA Status Not relevant AQIS Status Status not relevant

TGA Status Not relevant

Other References ADG Code - Australian Transport of Dangerous Goods

Workplace Exposure Standard for Airborne Contaminants

Approved Criteria for Classifying Hazardous Substances NOHSC: 1008 (2004)

Hazardous Substances Information System (HSIS)

Model Work Health and Safety Regulations 2011, Chapter 7 Hazardous Chemicals, Part 7.2 Lead.

Labelling of Workplace Hazardous Chemicals- Code Of Practice

Preparation of Safety Data Sheets for Hazardous Chemicals- Code of Practice

16. OTHER RELEVANT INFORMATION

Revision Information	Revision No	Date	Description
	1	28 /10/15	Initial SDS creation

Abbreviations AICS Australia Inventory of Chemical Substances

APVMA Australian Pesticides and Veterinary Medicines Authority

AQIS Australian Quarantine and Inspection Service

CAS # Chemical Abstract Service Number – used to uniquely identify chemical compounds

IARC International Agency for Research on Cancer

LC50 Lethal Concentration- toxicity of the surrounding medium that will kill half of the sample population of a specific test-

animal in a specified period through exposure via inhalation (respiration)

SDS Safety Data Sheet- (SDS), previously called a Material Safety Data Sheet (SDS),

TGA Therapeutic Goods Administration