1. PRODUCT IDENTIFICATION

Product Name: LEAD OXIDE
Other Names: PbO, lead oxide (PbO), lead containing, lead monoxide, Litharge
Use: Lead oxide – a raw material for the production of “lead paste” for battery manufacture
Supplier Name and Address: Century Yuasa Batteries
37-65 Cobalt St
Carole Park
QLD 4300
Telephone: (07) 3361 6161
Emergency (24 Hours): (07) 3361 6707

2. HAZARD(S) IDENTIFICATION

HAZARDOUS CHEMICAL DANGEROUS GOODS. According to the Model WHS Regulations and the ADG Code.

Poisons Schedule: S6 Classified as S6: Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP)

Signal Word: DANGER

GHS Classification: Acute Toxicity (Oral) Category 4, Acute Toxicity (Inhalation) Category 4, Reproductive Toxicity Category 1B, STOT - RE Category 2, Acute Aquatic Hazard Category 1, Chronic Aquatic Hazard Category 1

GHS Label Elements:
- Hazard Statements:
  - H302 Harmful if swallowed
  - H332 Harmful if inhaled
  - H360 May damage fertility or the unborn child
  - H373 May cause damage to organs through prolonged or repeated exposure
  - H400 Very toxic to aquatic life
  - H410 Very toxic to aquatic life with long lasting effects

Precautionary Statements:
- Prevention
  - P101 If medical advice is needed, have product container or label at hand.
  - P102 Keep out of reach of children
  - P103 Read label before use.
  - P201 Obtain special instructions before use.
  - P260 Do not breathe dust / fume / gas / mist / vapours / spray.
  - P270 Do not eat, drink or smoke when using this product.
- Response
  - P308+P313 IF EXPOSED: or concerned:
    - Get medical advice / attention.
  - P314 Get medical advice / attention if you feel unwell.
  - P301+P312 IF SWALLOWED: Call a POISON CENTER / doctor / physician / first aider / if you feel unwell.
  - P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.
  - P391 Collect spillage.
- Storage
  - P405 Store locked up.
- Disposal
  - P501 Dispose of contents, container to authorised chemical landfill or if organic, to high temperature incineration.
3. COMPOSITION, INFORMATION ON INGREDIENTS

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Identification</th>
<th>Content % weight</th>
</tr>
</thead>
<tbody>
<tr>
<td>Lead Monoxide (PbO)</td>
<td>CAS 1317-36-8</td>
<td>85%</td>
</tr>
<tr>
<td>Lead Oxide (PbO₂)</td>
<td>CAS 1309-60-0</td>
<td>15%</td>
</tr>
</tbody>
</table>

4. FIRST AID MEASURES

DESCRIPTION OF FIRST AID MEASURES

Eye contact
- If this product comes in contact with the eyes:
  - Wash out immediately with fresh running water.
  - Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
  - Seek medical attention without delay; if pain persists or recurs seek medical attention.
  - Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.

Skin contact
- If skin contact occurs:
  - Immediately remove all contaminated clothing, including footwear.
  - Flush skin and hair with running water (and soap if available).
  - Seek medical attention in event of irritation.

Inhalation
- If fumes or combustion products are inhaled:
  - Remove from contaminated area.
  - Lay patient down. Keep warm and rested.
  - Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.
  - Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.
  - Transport to hospital, or doctor.

Ingestion
- IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.
  - For advice, contact a Poisons Information Centre or a doctor.
  - Urgent hospital treatment is likely to be needed.
  - In the meantime, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient’s condition.
  - If the services of a medical officer or medical doctor are readily available, the patient should be placed in his / her care and a copy of the SDS should be provided. Further action will be the responsibility of the medical specialist.
  - If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the SDS.
  - Where medical attention is not immediately available or where the patient is more than 15 minutes from a hospital or unless instructed otherwise:
    - INDUCE vomiting with fingers down the back of the throat, ONLY IF CONSCIOUS. Lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.
    - NOTE: Wear a protective glove when inducing vomiting by mechanical means.

MEDICAL ATTENTION AND SPECIAL TREATMENT. Indication of any immediate medical attention and special treatment needed

Treat symptomatically.
- Gastric acids solubilise lead and its salts and lead absorption occurs in the small bowel.
- Particles of less than 1 um diameter are substantially absorbed by the alveoli following inhalation.
- Lead is distributed to the red blood cells and has a half-life of 35 days. It is subsequently redistributed to soft tissue & bone-stores or eliminated. The kidney accounts for 75% of daily lead loss; integumentary and alimentary losses account for the remainder.
- Neuropsychiatric symptoms are the most common symptoms of intoxication. Lead toxicity produces a classic motor neuropathy. Acute encephalopathy appears infrequently in adults. Diazepam is the best drug for seizures.
- Whole-blood lead is the best measure of recent exposure; free erythrocyte protoporphyrin (FEP) provides the best screening for chronic exposure. Obvious clinical symptoms occur in adults when whole-blood lead exceeds 80 ug / dL.
- British anti-lewisite is an effective antidote and enhances faecal and urinary excretion of lead. The onset of action of BAL is about 30 minutes and most of the chelated metal complex is excreted in 4-6 hours, primarily in the bile. Adverse reaction appears in up to 50% of patients given BAL in doses exceeding 5 mg / kg. CaNa₂EDTA has also been used alone or in concert with BAL as an antidote. D-penicillamine is the usual oral agent for mobilisation of bone lead; its use in the treatment of lead poisoning remains investigational. 2,3-dimercapto-1-propanesulphonic acid (DMPS) and dimercaptosuccinic acid (DMSA) are water soluble analogues of BAL and their effectiveness is undergoing review. As a rule, stop BAL if lead decreases below 50 ug / dL; stop; CaNa₂EDTA if blood lead decreases below 40 ug / dL or urinary lead drops below 2 mg / 24hrs.
5. FIRE FIGHTING MEASURES

Recommended Extinguishing Media
- Water spray or fog
- Foam
- Dry chemical powder
- Carbon dioxide
- BCF\ Vaporising Liquid

Extinguishing Media Incompatibilities
- There is no restriction on the type of extinguisher which may be used.
- Use extinguishing media suitable for surrounding area.

Specific Hazards Hazardous Decomposition
- Non-combustible.
- Not considered a significant fire risk, however containers may burn. Decomposition may produce toxic fumes of metal oxides, which may emit poisonous fumes

Fire Incompatibility
- None known.

Fire Fighting, Special Protective Equipment & Precautions
- Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves.
- Prevent, by any means available, spillage from entering drains or water courses.
- Use fire fighting procedures suitable for surrounding area.
- DO NOT approach containers suspected to be hot.
- Cool fire exposed containers with water spray from a protected location.
- If safe to do so, remove containers from path of fire.
- Equipment should be thoroughly decontaminated after use.

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions
- Avoid contact with skin and eyes.

Environmental Precautions
- Prevent, by any means available, spillage from entering drains or water course.

Methods and materials for containment and cleaning up
- With a clean shovel, transfer spilled material into clean-labelled containers for disposal.
- Prevent from entering drains, sewers, streams or other bodies of water. If contamination of sewers or waterways has occurred, advise the local emergency services

Protective Equipment
- Personal Protective Equipment advice is contained in Section 8 of the SDS.

Emergency Procedures

Minor Spills
- Clean up all spills immediately.
- Avoid contact with skin and eyes.
- Control personal contact with the substance, by using protective equipment.
- Use dry clean up procedures and avoid generating dust.
- Place in a suitable, labelled container for waste disposal.

Major Spills
- Remove all ignition sources.
- Clean up all spills immediately.
- Avoid contact with skin and eyes.
- Control personal contact with the substance, by using protective equipment.
- Clear area of personnel and move upwind.
- Alert Fire Brigade and tell them location and nature of hazard.
- Wear breathing apparatus plus protective gloves.
- Prevent, by any means available, spillage from entering drains or water courses.
- Stop leak if safe to do so.
- Contain or absorb spill with sand, earth or vermiculite.
- Collect recoverable product into labelled containers for recycling.
- Collect solid residues and seal in labelled drums for disposal.
- Wash area and prevent runoff into drains.
- After clean-up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.
- If contamination of drains or waterways occurs, advise emergency services, other related regulatory authorities such as environmental protection and local council.

7. HANDLING AND STORAGE

Safe Handling
- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- DO NOT allow material to contact humans, exposed food or food utensils.
- When handling, DO NOT eat, drink or smoke.
- Keep containers securely sealed when not in use.
- Avoid physical damage to containers.
- Always wash hands with soap and water after handling.
- Work clothes should be laundered separately. Launder contaminated clothing before re-use.
- Use good occupational work practice.
SAFETY DATA SHEET
LEAD OXIDE

• Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.

Storage
• Store in original containers.
• Keep containers securely sealed.
• Store in a cool, dry, well-ventilated area.
• Store away from incompatible materials and foodstuff containers.
• Protect containers against physical damage and check regularly for leaks.

Suitable container
• Packing of Lead oxide product in light weight metal or plastic packages may result in container collapse with product release
• Lined metal can, lined metal pail/can.
• Plastic pail.
• Poly-lined drum.
• Check all containers are clearly labelled and free from leaks.
• For materials with a viscosity of at least 2680 cSt (23 deg. C) and solids (between 15 C deg. and 40 deg C):
  • Removable head packaging;

Storage incompatibility
• Is incompatible with aluminium carbide, barium sulphide, silicon, sulphuryl chloride, hydrogen peroxide, chemical active metals, aluminium, combustible materials, lithium carbide, chlorinated rubber, chlorine, boron, hydrides, ethylene, fluorine, sulphides, acetylides and strong reducing agents.

FLAMMABLES
EXPLOSIVES
ACUTE TOXIC
OXIDISERS
HARMFUL
IRRITANT
CORROSIVE

8. EXPOSURE CONTROLS, PERSONAL PROTECTION

AUSTRALIAN EXPOSURE STANDARDS (Occupational Exposure Limits)

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Material name</th>
<th>TWA</th>
<th>STEL</th>
</tr>
</thead>
<tbody>
<tr>
<td>Lead Monoxide (PbO)</td>
<td>Lead, inorganic dusts &amp; fumes (as Pb)</td>
<td>0.15 mg / m3</td>
<td>Not Available</td>
</tr>
<tr>
<td>Lead oxide (PbO₂)</td>
<td>Lead, inorganic dusts &amp; fumes (as Pb)</td>
<td>0.15 mg / m3</td>
<td>Not Available</td>
</tr>
</tbody>
</table>

APPROPRIATE ENGINEERING CONTROLS

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:
• Process controls which involve changing the way a job activity or process is done to reduce the risk.
• Enclosure and/or isolation of emission source which keeps a selected hazard “physically” away from the worker and ventilation that strategically “adds” and “removes” air in the work environment.

PERSONAL PROTECTION

Respirator Type

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the “Exposure Standard” (or ES), respiratory protection is required.

• Type E-P Filter of sufficient capacity.

<table>
<thead>
<tr>
<th>Required Minimum Protection Factor</th>
<th>Half-Face Respirator</th>
<th>Full-Face Respirator</th>
<th>Powered Air Respirator</th>
</tr>
</thead>
<tbody>
<tr>
<td>up to 10 x ES</td>
<td>P1 Air-line*</td>
<td>-</td>
<td>PAPR-P1</td>
</tr>
<tr>
<td>up to 50 x ES</td>
<td>Air-line*</td>
<td>P2</td>
<td>PAPR-P2</td>
</tr>
<tr>
<td>up to 100 x ES</td>
<td>P3</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>100+ x ES</td>
<td>Air-line**</td>
<td>-</td>
<td>PAPR-P3</td>
</tr>
</tbody>
</table>

* Negative pressure demand ** Continuous flow

Eye Protection
• Safety glasses with side shields
• Chemical goggles.

Glove Type
• Wear chemical protective gloves, e.g. PVC

Clothing
• Overalls.

Footwear
• Wear safety footwear or safety gumboots e.g. Rubber

9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance: Pale yellow/bright yellow orange coloured, very dense, odourless powder. Does not mix with water but classed as “soluble” under the provisions of SP 199 of the UN Transport Code.

Odour: Not Available
Vapour pressure (kPa): 1 mm Hg @ 973 °C

Odour threshold: Not Available
Vapour density (Air = 1): Not Applicable
SAFETY DATA SHEET
LEAD OXIDE

pH
Not Applicable

Relative density (Water = 1)
9.53-9.6

Solubility in water (g/L)
Immisible

Initial boiling point and boiling range
(°C)
888 °C

1472 °C

Partition coefficient: n-octanol / water
Not Available

Flash point
Not Applicable

Molecular weight (g / mol)
223.2 g / mol

Evaporation rate
Not Available

Decomposition temperature (°C)
>500-700 °C lead fumes given off

Flammability
Not Applicable

Viscosity
Not Available

Upper, lower flammability or explosive limits
Not Applicable

10. STABILITY AND REACTIVITY

Reactivity
See section 7 and this section under Chemical stability

• Lead oxide:- is a strong oxidiser
• Attacks some plastics, rubber and coatings

Possibility of hazardous reactions
See section 5 & 7

• Reacts explosively with 90% performic acid, rubidium acetylde
• Reacts violently with strong oxidisers,
• Reacts violently with aluminium, sodium, zirconium, titanium, boron or silicon, when heated forms impact sensitive explosive mixtures with dichloromethylsilane

Incompatible materials
See section 7

• Is incompatible with aluminium carbide, barium sulphide, silicon, sulphuryl chloride, hydrogen peroxide, chemical active metals, aluminium, combustible materials, lithium carbide, chlorinated rubber, chlorine, boron, hydrides, ethylene, fluorine, sulphides, acetylides and strong reducing agents.

Chemical stability

• Product is considered stable
• Hazardous polymerisation will not occur.
• Unstable in the presence of incompatible materials

Hazardous decomposition products
Thermal decomposition may produce oxides of lead.

11. TOXICOLOGICAL INFORMATION ACUTE EFFECTS

No adverse health effects expected if the product is handled in accordance with this safety Data sheet and the product Label. Symptoms or effects that may arise if the product is mishandled and overexposure occurs are:-

Inhaled
The material is not thought to produce either adverse health effects or irritation of the respiratory tract following inhalation (as classified by EC Directives using animal models). Nevertheless inhalation of dusts, or fumes, especially for prolonged periods, may produce respiratory discomfort and occasionally, distress. Adverse systemic effects have been produced following exposure of animals by at least one other route and good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting. Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.

Inhalation of dusts, generated by the material, during the course of normal handling, may be harmful.

If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further risk if handling and use of the material result in excessive exposures.

Ingestion
Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual.

Skin contact
The material is not thought to be a skin irritant (as classified by EC Directives using animal models). Abrasive damage however, may result from prolonged exposures.

Skin contact with the material may damage the health of the individual; systemic effects may result following absorption.

Open cuts, abraded or irritated skin should not be exposed to this material.

Entry into the blood-stream, through for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects.

Eye
Although the material is not thought to be an irritant (as classified by EC Directives), direct contact with the eye may cause transient discomfort characterised by tearing or conjunctival redness (as with windburn). Slight abrasive damage may also result.

Chronic effects
An inorganic compound such as Lead is a cumulative harmful poison when exposed in small amounts can raise the body's content to toxic levels. Prolonged or repeated exposure to lead toxicity effects the nervous system (memory loss, tiredness, headaches, fatigue, irritability, decreased libido, dizziness, depression, encephalopathy (brain damage caused by altered brain function and structure), behavioural effects, altered mood states, disturbances in hand-eye coordination, reaction times, visual motor performance, and mental performance, disturbances to vision, changes in hearing, muscle and joint weakness of the arms and legs, (foot-drop and wrist-drop), heart / blood vessels (reduced haemoglobin synthesis and production, reduced life span and function of red blood cells, anaemia, increased blood pressure), digestive system (loss of appetite, anorexia, with severe abdominal pain, diarrhoea, inflammation of the stomach walls (gastritis) and colic, cramps, nausea, vomiting, constipation, weight loss and decreased urination, deposition of blue lead-line on the gums), kidneys / urinary system (reversible / irreversible kidney damage) and
endocrine system. Increased levels of lead result in increased brain damage, coma and death in extreme cases.

- Long term exposure to high dust concentrations may cause changes in lung function i.e. pneumoconiosis; caused by particles less than 0.5 micron penetrating and remaining in the lung.
- Ample evidence from experiments exists that there is a suspicion this material directly reduces fertility.
- Lead can cross the placenta, and cause miscarriage, stillbirths and birth defects. Exposure before birth can cause mental retardation, behavioural disorders and infant death.
- Exposure to the material for prolonged periods may cause physical defects in the developing embryo (teratogenesis).
- Ample evidence exists that developmental disorders are directly caused by human exposure to the material.
- Lead can accumulate in the skeleton for a very long time.

### Acute Toxicity

<table>
<thead>
<tr>
<th>Condition</th>
<th>Skin Irritation / Corrosion</th>
<th>Serious Eye Damage / Irritation</th>
<th>Respiratory Or Skin Sensitisation</th>
<th>Mutagenicity</th>
<th>Carcinogenicity</th>
<th>Reproductivity</th>
<th>Stot - Single Exposure</th>
<th>Stot - Repeated Exposure</th>
<th>Aspiration Hazard</th>
</tr>
</thead>
<tbody>
<tr>
<td>✔</td>
<td></td>
<td></td>
<td></td>
<td></td>
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<td></td>
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<td></td>
</tr>
</tbody>
</table>

**For Lead:**

- **Environmental Fate:** Lead is assessed as low hazard if it remains in its solid, massive, metallic form. Lead, in the form of alkyls, has been introduced to the environment primarily from leaded gasoline / petrol. These are converted to water-soluble lead compounds of high toxicity and availability to plants.
- **Atmospheric Fate:** Lead is primarily an atmospheric pollutant that enters soil and water as fallout, a process determined by the physical form involved and particle size. Lead, in the form of alkyls, has been introduced to the environment primarily from leaded gasoline / petrol. Lead is absorbed by mammals / humans via vapors, contaminated dust, and fumes.
- **Terrestrial Fate:** Soil - Lead alkyls easily leach from soil to contaminate water sources close to highways. Plants - Lead alkyls that have been converted to water soluble lead compounds have high toxicity / availability to plants.
- **Aquatic Fate:** Lead that has entered the aquatic system is expected to be found in sediments.
- **Ecotoxicity:** Soluble or insoluble lead may enter the environment and accumulate. Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

**Fish**

The following applies to lead compounds in general:
- Fish: lethal from 1.4 mg / l / up S. gairdneri: LC50: 0.14 mg / l / 96h L. idus LC50: 546 mg / l fish test LC50: 236 mg / l (calc. as free lead).

**Daphnia**

The following applies to lead compounds in general:
- Fish: lethal from 1.4 mg / l / up S. gairdneri: LC50: 0.14 mg / l / 96h L. idus LC50: 546 mg / l fish test LC50: 236 mg / l (calc. as free lead).

**Algae**

The following applies to lead compounds in general:
- Algae: Sc. quadracuda toxic from 3.7 mg / l / up M. aeruginosa 0.45 mg / l (calc. as free lead).

**Bacteria**

The following applies to lead compounds in general:
- Algae: Sc. quadracuda toxic from 3.7 mg / l / up M. aeruginosa 0.45 mg / l (calc. as free lead).

**Other Organisms**

The following applies to lead compounds in general:
- Protozoa: E. sulcatum toxic from 0.02 mg / l / up U. parduczi toxic from 0.07 mg / l / up (calc. as free lead).

**Degradability**

No Data available for all ingredients

**Bio-accumulative Potential**

- Lead Monoxide LOW (BCF = 43)

**Mobility in Soil**

No Data available for all ingredients

**Other Adverse Effects**

No Data available for all ingredients

### Disposal Considerations

- Dispose in accordance with federal, state or local regulations.
- Containers may still present a chemical hazard / danger when empty.
- Return to supplier for reuse / recycling if possible.
- Otherwise:
  - If container cannot be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
  - Where possible retain label warnings and SDS and observe all notices pertaining to the product.
  - Legislation addressing waste disposal requirements may differ by country, state and / or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.
  - This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. In most instances the material is still present as a chemical hazard / danger when empty.

### 12. ECOLOGICAL INFORMATION

**Toxicity**

- DO NOT discharge into sewer or waterways.
- Very toxic to aquatic organisms. May cause long-term adverse effects in the aquatic environment.
- DO NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.
- Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

**For Lead:**

- **Environmental Fate:** Lead is assessed as low hazard if it remains in its solid, massive, metallic form. Lead, in the form of alkyls, has been introduced to the environment primarily from leaded gasoline / petrol. These are converted to water-soluble lead compounds of high toxicity and availability to plants.
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- Protozoa: E. sulcatum toxic from 0.02 mg / l / up U. parduczi toxic from 0.07 mg / l / up (calc. as free lead).

**Degradability**

No Data available for all ingredients

**Bio-accumulative Potential**

- Lead Monoxide LOW (BCF = 43)

**Mobility in Soil**

No Data available for all ingredients

**Other Adverse Effects**

No Data available for all ingredients
supervisor of the material should be consulted.

• DO NOT allow wash water from cleaning or process equipment to enter drains.
• It may be necessary to collect all wash water for treatment before disposal.
• In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
• Where in doubt contact the responsible authority.
• Recycle wherever possible or consult manufacturer for recycling options.
• Consult State Land Waste Management Authority for disposal.
• Bury residue in an authorised landfill.
• Recycle containers if possible, or dispose of in an authorised landfill.
• Observe all label safeguards until containers are cleaned and destroyed.

Environmental Regulations

• Dispose in accordance with federal, state or local regulations.
• Refer to section 15

14. TRANSPORT INFORMATION

REGULATED FOR TRANSPORT OF DANGEROUS GOODS ADG

UN Number 2291
Proper Shipping Name LEAD COMPOUND, SOLUBLE, N.O.S.
Transport Hazard Class Class: 6.1 Sub risk: Not Applicable
Packing group III
Environmental Hazards No relevant data
Special Precautions Special provisions 199, 274
Limited quantity 5 kg
Additional Information Marine Pollutant: Yes
Hazchem Code 2Z

15. REGULATORY INFORMATION

SAFETY, HEALTH AND ENVIRONMENTAL REGULATIONS, LEGISLATION

Lead Monoxide (PbO) Lead Oxide (PbO2) are found on the following regulatory lists


Poisons Schedule (Australia) 6 AICS Status All the constituents of this product are listed
APVMA Status Not relevant AQIS Status Status not relevant
TGA Status Not relevant

Other References ADG Code - Australian Transport of Dangerous Goods
Workplace Exposure Standard for Airborne Contaminants
Hazardous Substances Information System (HSIS)
Model Work Health and Safety Regulations 2011, Chapter 7 Hazardous Chemicals, Part 7.2 Lead.
Labelling of Workplace Hazardous Chemicals- Code Of Practice
Preparation of Safety Data Sheets for Hazardous Chemicals- Code of Practice

16. OTHER RELEVANT INFORMATION

Revision Information

<table>
<thead>
<tr>
<th>Revision No</th>
<th>Date</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>28/10/15</td>
<td>Initial SDS creation</td>
</tr>
</tbody>
</table>

Abbreviations

AICS Australia Inventory of Chemical Substances
APVMA Australian Pesticides and Veterinary Medicines Authority
AQIS Australian Quarantine and Inspection Service
CAS # Chemical Abstract Service Number – used to uniquely identify chemical compounds
IARC International Agency for Research on Cancer
LC50 Lethal Concentration- toxicity of the surrounding medium that will kill half of the sample population of a specific test-animal in a specified period through exposure via inhalation (respiration)
SDS Safety Data Sheet- (SDS), previously called a Material Safety Data Sheet (SDS).
TGA Therapeutic Goods Administration